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Jan Ramza^a; Gerard Descotes^a; Jean-Marie Basset^b; Andy Mutch^b

^a Université Claude Bernard Lyon I, Laboratoire Chimie Organique II, Villeurbanne, Cedex, France ^b Laboratoire de Chimie Organométallique de Surface, Villeurbanne, Cedex, France

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Metathesis of Ω -Unsaturated Glycosides with Chloro-Aryloxo Complexes of Tungsten, as a New Synthetic Route Leading to Unsaturated Neutral Bolaforms

Jan Ramza,^a Gerard Descotes,^a Jean-Marie Basset,^b and Andy Mutch^b

^aUniversité Claude Bernard Lyon I, Laboratoire Chimie Organique II, URA CNRS 463
43, Bd. du 11 Novembre 1918, 69622 Villeurbanne, Cedex, France

^bLaboratoire de Chimie Organométallique de Surface, CNRS UMR 9986, 2 rue Albert
Einstein 69626 Villeurbanne, Cedex, France

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ABSTRACT

Glucosides of unsaturated alcohols, bearing different functional groups on the carbohydrate units, underwent catalytic metathesis reaction affording neutral bolaforms in good yields. Tungsten aryloxo-complexes (**1** and **2**), due to large steric hindrance, were found to be superior catalysts for this process.

INTRODUCTION

Bolaforms represent a group of compounds which consist of two hydrophilic heads joined together by a long hydrophobic spacer (aliphatic chain). During recent years interest in this group of compounds has grown, because of their unusual properties as surfactants^{1,2} (they are able to form micelles and vesicles), and their ability to form very thin monolayer lipid membranes.³ Bolaforms have also been used for studying mechanisms of membrane processes,² as well as some enzymatic reactions,^{4,5} and have

been investigated as liquid crystals.⁶ Consequently, the practical use of these compounds is expected to grow in the near future. Among one of the most interesting classes of bolaforms is the neutral form containing carbohydrate units at both ends of the long aliphatic chain.

As previously reported,⁷ one of the most simple and effective synthetic routes leading to carbohydrate-based bolaforms is the olefin metathesis - a well known catalytic reaction intensively studied for thirty years, because of numerous practical applications.^{8,9,10} This reaction was mainly applied to acyclic and cyclic olefins, and in a few cases to olefins bearing single functional groups^{11,12} (e.g., esters, ethers, thioethers, quaternary ammonium salts, nitriles). Application of this catalytic reaction for polyfunctional, polyoxygenated substrates, as carbohydrates has extended the scope of synthetic utility of this method. Here we present full experimental details for our preliminary results in this area, and an extension of the investigation employing glucoside substrates bearing more than one olefinic function.

RESULTS AND DISCUSSION

Recently developed by us, a new generation of chloro-aryloxide complexes of tungsten, (**1** and **2**, Figure 1) are among the most active catalysts in metathesis of olefins bearing functional groups (e.g., esters, thioethers, phosphoranes).^{13,14} Catalyst **1** is a bicomponent system, associating an air stable chloro-aryloxide tungsten (VI) complex with an alkylating agent such as SnR₄ and preferably PbR₄ (R = Et, Bu). It is considered, that the alkylating agent is necessary to carry out a double alkylation of the tungsten, followed by α -hydrogen transfer to generate *in situ* the metallocarbene. The most active and stereoselective catalyst **2** is derived from **1**, a cyclo metallated aryloxo (chloro) neopentylidene tungsten (VI) complex.^{12,15}

Since the ligand environment around the tungsten atom is an extremely bulky (*O*, *O'*- diphenylphenoxy) group, and since bulky groups have difficulty coordinating to the metallocarbene, these catalysts are expected to tolerate a wide range of functional groups, especially esters and ethers. So it was logical to test these complexes with unsaturated *O*-protected carbohydrates, namely ω -unsaturated *O*-protected glucosides. The use of protecting groups on the sugar residue would seem to be necessary to prevent deactivation of the oxophilic and moisture sensitive catalysts with a *d*^o configuration. Some of the most commonly used *O*-protecting groups, namely acetyl, *t*-butyldimethylsilyl and benzyl were selected for preliminary experiments. Classical glycosylation¹⁶ of 10-undecen-1-ol by 2,3,4,6-tetra-*O*-acetylglucopyranosyl bromide in the presence of the silver triflate, following deprotection and reprotection procedures^{17,18} afforded the desired ω -unsaturated substrates **3,4,5**.

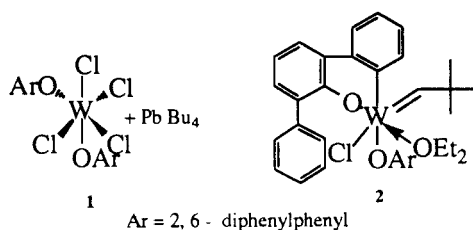


Figure 1

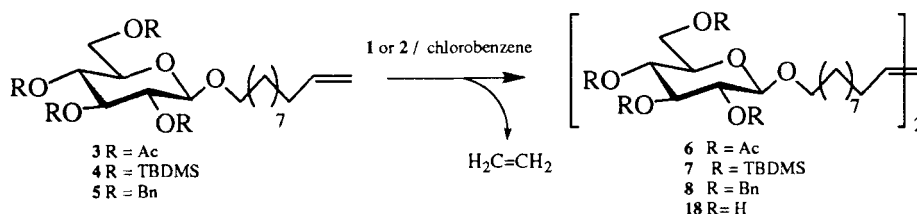


Figure 2

These substrates in the presence of catalysts **1** or **2** in chlorobenzene at 80 °C under argon gave expected metathesis products as shown in Figure 2.

Glucosides protected as acetyl esters or silyl ethers were found to be superior substrates in the metathesis reaction to the benzyl ether protected carbohydrate, the latter deactivating the catalyst and resulting in lower yield, even when a higher catalyst / substrate ratio was employed. (see Table 1).

The scope of this reaction was extended to *N*-acetyl aminosugars by successful application of glucosamine derivative **9** as another metathesis substrate (Figure 3).

4-Pentenyl and allyl glucosides **10** and **11**, prepared by the same glycosylation procedure as above, using 4-penten-1-ol and allyl alcohol, respectively, were also tested. Catalysts **1** and **2** were also active for the pentenyl glucoside **10** and results were as good as for the undecenyl substrate, whereas allyl glucoside **11** did not give any metathesis product. As we expected, in the case of allyl glucoside **11** as well as the benzyl protected substrate **5**, catalysts were deactivated by competitive coordination of the ether oxygen atoms to the metallocarbene. For silylated substrate **4**, the steric hindrance from the TBDMS group was large enough to inhibit this kind of complexation with the also sterically hindered tungsten. Similar steric effects were found to occur in metathesis of thioethers.¹⁹

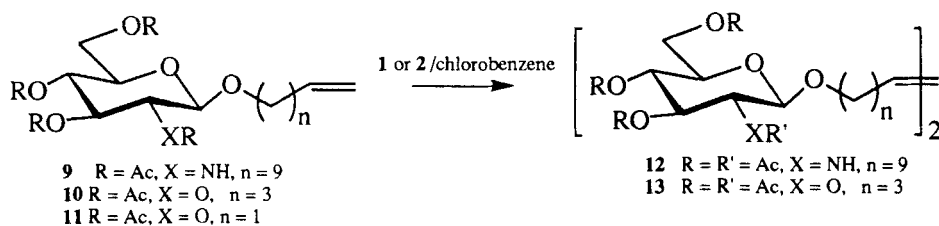


Figure 3

For the substrate **4** the catalyst developed by Mol¹¹ ($\text{Re}_2\text{O}_7 / \text{Al}_2\text{O}_3 - \text{SiO}_2 / \text{Pb Bu}_4$), gave no metathesis product. This result supports our hypothesis that steric hindrance effects play a key role in the metathesis reaction of polyfunctional compounds as the protection factor against catalyst deactivation.

Finally, we applied compounds **14** and **15** bearing two different olefinic groups as metathesis substrates. These substrates were prepared according to the method of Ferrier and Prasad²⁰ from 3,4,6-tri-*O*-acetyl-D-glucal with appropriate ω -unsaturated alcohols in the presence of BF_3 etherate. In both cases only the terminal isolated double bond was involved in the catalytic process (Figure 4). No sign of reaction involving the olefin part of the sugar ring was observed. We also did not observe any detectable increase in the deactivation rate of the catalyst by the ring olefin bearing three oxygen atoms at allylic positions.

Separations of products by column chromatography afforded for all metathesis reactions inseparable *E/Z* mixtures, but with predominance of one isomer. All reaction products were analysed by ^1H and ^{13}C NMR spectroscopy: proton and carbon atom signals of terminal double bond for substrates were replaced in products spectra by signals of symmetrically substituted olefins (results are summarized in Tables 1 and 2). However, it was not possible even after ^1H and ^{13}C NMR analysis to prove which isomer was the major reaction product.

Finally compound **6** was deprotected, by treatment with sodium methoxide in methanol, affording neutral bolaform **18**.

In conclusion, catalysts **1** and **2** appeared to be very efficient even in the presence of numerous functional groups. The steric hindrance from crowded aromatic substituents probably prohibits deactivation of the active metallocarbene by oxygen, and nitrogen atoms from functional groups of substrates. As result, metathesis was shown to be very effective for carbohydrate substrates, and an efficient method leading to unsaturated neutral bolaforms was developed.

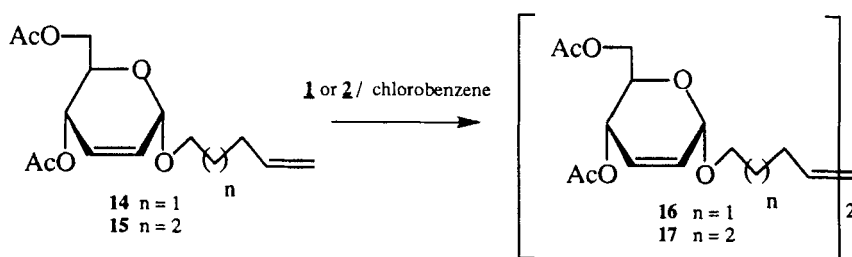


Figure 4

Table 1. Results obtained for catalytic metathesis reactions.

Substrate	Catalyst	Catalyst/substrate ratio	Reaction time hrs	Reaction temp. °C	Product	Yield %
3	2	1 : 10	12	80	6	64
3	1	1 : 5	12	80	6	52
4	1	1 : 12	12	65	7	51
4	2	1 : 12	12	65	7	92
5	1	1 : 5	12	80	8	37
5	2	1 : 15	12	80	no reaction	0
9	1	1 : 10	12	80	12	62
10	1	1 : 20	6	80	13	65
11	2	1 : 10	12	80	no reaction	0
11	1	1 : 8	12	80	no reaction	0
14	1	1 : 15	6	80	16	53
15	1	1 : 20	12	80	17	55

Table 2. Characteristic NMR data for olefinic protons and carbon atoms.

Substrates						Products		
N°	H α	H α'	H β	C α	C β	N°	H α =H β	C α =C β
3	5.08	4.87	5.82	114.4	139.3	6	5.38	130.5
4	5.06	4.92	5.81	113.8	139.5	7	5.37	130.4
5	4.99	4.91	5.82	114.2	139.4	8	5.39	130.3
9	5.07	4.90	5.81	114.7	139.6	12	5.36	130.3
10	5.02	4.85	5.71	115.0	137;7	13	5.38	129.0
14	5.10	4.85	5.10	115.0	138.1	16	5.41	129.0
15	5.10	4.90	5.85	114.4	140.6	17	5.39	129.5

EXPERIMENTAL

General methods. All reactions were performed under argon atmosphere in flame dried reactions flasks. All solvents were distilled prior to use. Chlorobenzene and methylene chloride were dried according to standard procedures.²¹ For column chromatography silica gel 60, 400-230 mesh (E. Merck) was employed. ¹H NMR and ¹³C NMR spectra were measured on a Bruker AM-200 spectrometer with TMS and CDCl₃ as internal standards respectively, with CDCl₃ as a solvent. For some cases decoupling experiments were done to ensure proton assignments. IR spectra were obtained using a Perkin Elmer 681 Infrared Spectrometer. Optical rotations were measured using a Perkin Elmer 241 Polarimeter.

10-Undecenyl 2,3,4,6-Tetra-O-acetyl-β-D-glucopyranoside 3. To a solution of 2,3,4,6-tetra-O-acetyl-glucopyranosyl bromide (5 g, 12.2 mM) and 10-undecen-1-ol (3.71 g, 12.8 mM) in CH₂Cl₂ (25 mL) dried molecular sieves were added and after the mixture was stirred for 45 min at rt AgOTf (3.28 g, 12.8 mM) was added at 0 °C in one portion. The mixture was stirred overnight at rt. Triethylamine (2.04 mL, 14.6 mM) was added and the mixture was filtered (celite). Chromatography of the oily residue after solvent evaporation (hexane/ethyl acetate 7:1) gave compound **3** (4.09 g, 67%) as an oil: [α]_D -17.4 (c 0.7, chloroform); IR ν_{max} 2920, 1750, 1640, 1225, 1040 cm⁻¹; ¹H NMR δ 5.80 (dddd, J_{10'11'}cis 10.2, J_{10'11'}trans 17.0, 2x J_{9',10'} 6.6, 1H, H-10'), 5.21 (dd, J_{3,4} 9.5, 1H, H-3), 5.12 (dd, J_{4,5} 9.6, 1H, H-4), 5.03 (dd, J_{2,3} 9.1, 1H, H-2), 4.97 (m, 2H, H-11'), 4.50 (d, J_{1,2} 7.9, 1H, H-1), 4.28 (dd, J_{5,6} 4.6, 1H, H-6), 4.14 (dd, J_{6,6'} 12.3, 1H, H-6'), 3.88 (ddd, J_{1',1''} 9.6, 2xJ_{1',2'} 6.2, 1H, H-1'), 3.70 (ddd, J_{5,6'} 2.4, 1H, H-5), 3.47 (ddd, 2xJ_{1',2'} 6.5, 1H, H-1''), 2.09, 2.04, 2.03, 2.01 (4xs, 12H, 4xCH₃CO), 2.0 (m, 2H H-9'), 1.56 (m, 2H, H-2'), 1.27 (m, 12H, CH₂ alkyl); ¹³C NMR 170.8, 170.4, 169.6, 169.4 (4 x CO), 139.3 (C-10'), 114.7 (C-11'), 101.0 (C-1), 73.0 (C-3), 71.9 (C-5), 71.5 (C-2), 70.4 (C-1'), 68.7 (C-4), 62.2 (C-6), 34.0-26.0 (8xC alkyl), 20.9-20.8 (4 x CH₃ CO).

Anal. Calcd for C₂₅H₄₀O₁₀: C, 59.98; H, 8.05. Found: C, 59.98; H, 8.25.

10-Undecenyl 2,3,4,6-Tetra-O-tert-butylidimethylsilyl-β-D-glucopyranoside 4. To the solution of compound **3** (1.0 g, 2.0 mM) in methanol (15 mL) K₂CO₃ (0.2 g) was added and the mixture was stirred 4 h at rt. Chromatography through a short column of silica gel (CH₂Cl₂/MeOH 4:1) of the oily residue after solvent evaporation gave deprotected compound (0.63 g, 95%); IR ν_{max} 3600, 2920, 1640, 1225, 1040 cm⁻¹, which was silylated by a standard procedure¹⁷ using imidazole (10 equiv) and *tert*-butyldimethylsilyl chloride (10 equiv). After purification compound **4** was obtained as colourless oil (76 %): [α]_D -16.1 (c 2.1, chloroform); IR ν_{max} 2930, 1640, 1220, 1030

cm^{-1} ; $^1\text{H NMR } \delta$ 5.82 (dddd, $J_{10'11'cis}$ 10.2, $J_{10'11'trans}$ 17.0, $2 \times J_{9',10'}$ 6.6, 1H, H-10'), 4.97 (m, 2H, H-11'), 4.67 (d, $J_{1,2}$ 6.7, 1H, H-1), 3.95-3.65 (m, 6H, H-3, 4, 5, 6, 6', 1'), 3.57 (dd, $J_{2,3} < 1.0$, 1H, H-2), 3.38 (ddd, $2 \times J_{1',2'}$ 6.5, $J_{1',1''}$ 8.9, 1H, H-1''), 2.0 (m, 2H H-9'), 1.62 (m, 2H, H-2'), 1.28 (m, 12H, CH_2 alkyl), 0.89-0.87 (4xs, 36H, $4 \times (\text{CH}_3)_3\text{CSi}$), 0.09-0.05 (8xs, 24H, $8 \times \text{CH}_3\text{Si}$); $^{13}\text{C NMR}$: 139.3 (C10'), 114.2 (C-11'), 101.8 (C-1), 82.4 (C-3), 79.3 (C-2), 77.9 (C-4), 70.2 (C-5), 69.6 (C-1'), 64.3 (C-6), 34.0-26.3 ($8 \times \text{C alkyl}$), 26.2-25.9 ($12 \times \text{CH}_3$ *t*-butyl), 18.4-18.0 ($4 \times \text{SiCMe}_3$), -4.0-5.1 ($8 \times \text{SiCH}_3$).

Anal. Calcd for $\text{C}_{41}\text{H}_{88}\text{O}_6$ Si₄: C, 62.38; H, 11.24. Found: C, 62.66; H, 11.30.

10-Undecenyl 2,3,4,6-Tetra-*O*-benzyl- β -D-glucopyranoside 5. Deacetylation of **3** was performed as above, and deprotected compound was benzylated by a standard procedure¹⁸ with NaH (6 equiv) and benzyl bromide (6 equiv). After purification, compound **5** was obtained as a colourless oil (63 %): $[\alpha]_{\text{D}} -6.7$ (*c* 0.08, chloroform); IR ν_{max} 3005, 2920, 1640, 1550, 1225, 1040 cm^{-1} ; $^1\text{H NMR } \delta$ 7.5-7.2 (m, 20H, arom), 5.88 (dddd, $J_{10'11'cis}$ 10.2, $J_{10'11'trans}$ 17.0, $2 \times J_{9',10'}$ 6.6, 1H, H-10'), 4.99 (m, 2H, H-11'), 4.85, 4.83, 4.68, 5.59 (4 x ABq, J_{AB} 11.4, 11.3, 10.8, 12.4, 8H, CH_2Ph), 4.40 (d, $J_{1,2}$ 7.7, 1H, H-1), 3.97 (ddd, $2 \times J_{1',2'}$ 6.3, $J_{1',1''}$ 9.3, 1H, H- α), 3.74 (dd, $J_{5,6}$ 2.0, $J_{6,6'}$ 10.5, 1H, H-6), 3.72-3.38 (m, 6H, H-2, 3, 4, 5, 6', 1''), 2.03 (m, 2H, H-9'), 1.62 (m, 2H, H-2'), 1.29 (m, 12H, CH_2 alkyl); $^{13}\text{C NMR}$: 139.4 (C-10'), 138.7-138.2 ($4 \times \text{C arom}$), 128.4-127.6 (C arom), 114.2 (C-11'), 103.7 (C-1), 84.7 (C-3), 82.3 (C-2), 78.0 (C-4), 75.7, 75.0, 74.7, 73.5 ($4 \times \text{CH}_2$ benzyl), 74.8 (C-5), 70.2 (C-1'), 69.0 (C-6), 32.6-26.2 (C alkyl).

Anal. Calcd for $\text{C}_{45}\text{H}_{56}\text{O}_6$: C, 78.00; H, 8.15. Found: C, 77.64; H, 7.91.

4-Pentenyl 2,3,4,6-Tetra-*O*-acetyl- β -D-glucopyranoside 10. This compound was prepared by the same procedure as **3** using 2,3,4,6-tetra-*O*-acetylglucopyranosyl bromide (5 g, 12.2 mM), 4-penten-1-ol (1.10 g, 12.8 mM) and AgOTf (3.28 g, 12.8 mM). Chromatography of the oily residue after solvent evaporation (hexane/ethyl acetate 7:1) gave compound **10** (3.21 g, 63%) as an oil $[\alpha]_{\text{D}} -13.2$ (*c* 1.4, chloroform); IR ν_{max} 2930, 1740, 1640, 1225, 1040 cm^{-1} ; $^1\text{H NMR } \delta$ 5.72 (dddd, $J_{4',5'cis}$ 10.3, $J_{4',5'trans}$ 17.0, $2 \times J_{3',4'}$ 6.7, 1H, H-4'), 5.15 (dd, $J_{3,4}$ 9.5, 1H, H-3), 5.04 (dd, $J_{4,5}$ 9.3, 1H, H-4), 4.93 (dd, $J_{2,3}$ 9.0, 1H, H-2), 4.90 (m, 2H, H-5'), 4.46 (d, $J_{1,2}$ 7.9, 1H, H-1), 4.22 (dd, $J_{5,6}$ 4.8, 1H, H-6), 4.08 (dd, $J_{6,6'}$ 12.3, 1H, H-6'), 3.82 (ddd, $J_{1',1''}$ 9.7, $2 \times J_{1',2'}$ 6.2, 1H, H-1'), 3.65 (ddd, $J_{5,6'}$ 2.4, 1H, H-5), 3.46 (ddd, $2 \times J_{1',2'}$ 6.5, 1H, H-1''), 2.03, 1.99, 1.97, 1.95 (4xs, 12H, $4 \times \text{CH}_3\text{CO}$), 2.0 (m, 2H H-3'), 1.62 (m, 2H, H-2'); $^{13}\text{C NMR}$: 170.5, 170.2, 169.3, 169.2 ($4 \times \text{CO}$), 137.7 (C-4'), 115.0 (C-5'), 100.7 (C-1), 72.8 (C-3), 71.7 (C-5), 71.3 (C-2), 69.2 (C-1'), 68.4 (C-4), 61.9 (C-6), 29.8, 28.5 ($2 \times \text{CH}_2$ alkyl), 20.6-20.5 ($4 \times \text{CH}_3\text{CO}$).

Anal. Calcd for C₁₉H₂₈O₁₀: C, 54.80; H, 6.78. Found: C, 54.35; H, 6.91.

4-Pentenyl 4,6-Di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno-pyranoside 14. To the solution of 3,4,6-tri-O-acetyl-D-glucal (1.08 g, 3.96 mM) in dry toluene (10 mL) 4-penten-1-ol (0.818 mL, 7.90 mM) was added, followed by boron trifluoride etherate (0.268 mL, 2.05 mM). The resulting mixture was stirred at rt for 3 h. An excess of aqueous NaHCO₃ was added, and after extraction, drying and solvent evaporation the product was purified by column chromatography (hexane/ethyl acetate 4:1) affording **14** (1.02 g, 82%) as a colourless oil: [α]_D +94.4 (*c* 5.0, chloroform); IR ν_{\max} 2900, 1750, 1640, 1225, 1030 cm⁻¹; ¹H NMR δ 5.83 (m, 2H, H-2, H-3), 5.82 (dddd, J_{4'5'}cis 10.2, J_{4'5'}trans 16.9, 2x J_{3'4'} 6.7, 1H, H-4'), 5.30 (dd, J_{5,4} 9.4, J_{3,4} 0.8, 1H, H-4), 5.02 (m, 2H, H-5'), 4.99 (bs, J_{1,2}=J_{1,3} 1.6, 1H, H-1), 4.20 (m, 2H, H-6, H-6'), 4.09 (ddd, J_{5,6} 2.4, J_{5,6'} 5.1, 1H, H-5), 3.79 (ddd, J_{1',1''} 9.6, 2xJ_{1',2'} 6.7, 1H, H-1'), 3.51 (ddd, 2xJ_{1'',2''} 6.5, 1H, H-1''), 2.15 (m, 2H H-3'), 2.09, 2.08 (2xs, 6H, 2xCH₃CO), 1.71 (m, 2H, H-2'); ¹³C NMR: 170.8, 170.3 (2x CO), 138.1 (C-4'), 130.6 (C-3), 126.2 (C-2), 114.9 (C-5'), 95.3 (C-1), 72.8 (C-5), 67.8 (C-1'), 64.5 (C-4), 63.5 (C-6), 30.3, 28.9 (2xCH₂ alkyl), 21.0, 20.9 (2 x CH₃ CO).

Anal. Calcd for C₁₅H₂₂O₆: C, 60.39; H, 7.43. Found: C, 60.48; H, 7.38.

5-Hexenyl 4,6-Di-O-acetyl-2,3-dideoxy- α -D-erythro-hex-2-eno-pyranoside 15. This compound was prepared by the same procedure as for **14**. Purified product **15** was obtained in 60% yield as a colourless oil: [α]_D +61.2 (*c* 2.5, chloroform); IR ν_{\max} 2910, 1750, 1640, 1220, 1040 cm⁻¹; ¹H NMR δ 5.86 (m, 2H, H-2, H-3), 5.80 (dddd, J_{6'5'}cis 10.2, J_{6'5'}trans 16.9, 2x J_{4',5'} 6.7, 1H, H-5'), 5.31 (dd, J_{5,4} 9.5, J_{3,4} 0.8, 1H, H-4), 5.03 (bs, J_{1,2}=J_{1,3} 1.6, 1H, H-1), 4.96 (m, 2H, H-6'), 4.24 (dd, J_{5,6} 4.9, 1H, H-6), 4.17 (dd, J_{6,6'} 12.3, J_{5,6'} 2.4, 1H, H-6'), 4.09 (ddd, 1H, H-5), 3.78 (ddd, J_{1',1''} 9.7, 2xJ_{1',2'} 6.7, 1H, H-1'), 3.52 (ddd, 1H, H-1''), 2.10, 2.08 (2xs, 6H, 2xCH₃CO), 2.07 (m, 2H H-4'), 1.63 (m, 2H, H-2'), 1.48 (m, 2H, H-3'); ¹³C NMR: 170.0, 170.0 (2x CO), 140.6 (C-5'), 129.2 (C-3), 126.8 (C-2), 114.4 (C-6'), 102.0 (C-1), 77.8 (C-5), 73.0 (C-4), 65.9 (C-6), 64.5 (C-1'), 33.5, 31.5, 26.3 (3xCH₂ alkyl), 17.7, 17.3 (2 x CH₃ CO).

General procedures for the metathesis reaction:

a. with the bicomponent catalyst **1**; to a solution of tetrachloro-bis-(2,6-diphenylphenoxy)tungsten **1** (8.2 mg, 0.01 mM) in dry chlorobenzene (2 mL) 2 equiv of PbBu₄ (6.9 μ L) were added by syringe and the solution was stirred 15 min at 80 °C. The unsaturated substrate (5-20 equiv) was added at the same temperature, and stirring was continued overnight. Products were separated by column chromatography after evaporation of solvent under reduced pressure.

b. with the monocomponent catalyst **2**; to a solution of complex **2** (8.5 mg, 0.01 mM) in dry chlorobenzene (2 mL) the unsaturated substrate (5-20 equiv) was added at 80 °C, and the mixture was stirred overnight. Products were separated by column chromatography after evaporation of solvent under reduced pressure.

(E,Z)-Di-O-(2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyloxy)-1,20-eicos-10-ene 6. Product obtained from substrate **3** by procedure **a**, (catalyst / substrate ratio - 1:5), yield: 52%, as an oil: $[\alpha]_D -1.78$ (*c* 0.23, chloroform); IR ν_{\max} 2920, 1750, 1640, 1225, 1040 cm^{-1} ; $^1\text{H NMR}$ δ 5.38 (m, 1H, H-10'), 5.19 (dd, $J_{3,4}$ 9.3, 1H, H-3), 5.10 (dd, $J_{4,5}$ 9.5, 1H, H-4), 4.99 (dd, $J_{2,3}$ 9.3, 1H, H-2), 4.49 (d, $J_{1,2}$ 7.9, 1H, H-1), 4.27 (dd, $J_{5,6}$ 4.6, 1H, H-6), 4.15 (dd, $J_{6,6'}$ 12.2, 1H, H-6'), 3.86 (ddd, $J_{1',1''}$ 9.6, $2 \times J_{1',2'}$ 6.4, 1H, H-1'), 3.69 (ddd, $J_{5,6'}$ 2.4, 1H, H-5), 3.46 (ddd, $2 \times J_{1'',2'}$ 6.4, 1H, H-1''), 2.09, 2.04, 2.03, 2.01 (4xs, 12H, 4xCH₃CO), 1.97 (m, 2H H-9'), 1.57 (m, 2H, H-2'), 1.26 (m, 12H, CH₂ alkyl); $^{13}\text{C NMR}$: 170.9, 170.6, 169.8, 169.5 (4 x CO), 130.6, 130.1 (*cis* and *trans* C-10'), 101.0 (C-1), 73.1 (C-3), 71.9 (C-5), 71.5 (C-2), 70.5 (C-1'), 68.7 (C-4), 62.2 (C-6), 32.8-26.0 (8xC alkyl), 21.0-20.9 (4 x CH₃ CO). FABMS: 996.0 (M+Na)⁺, 990.9 (M+NH₄)⁺, 523.5, 331.4.

Anal. Calcd for C₄₈H₇₆O₂₀: C, 59.25; H, 7.87. Found: C, 58.98; H, 7.85.

The same product **6** was obtained by procedure **b**, (catalyst / substrate ratio - 1:10), yield: 64%.

(E,Z) Di-O-(2,3,4,6-tetra-O-tert-butyltrimethylsilyl- β -D-glucopyranosyloxy)-1,20-eicos-10-ene 7. The product was obtained from substrate **4** by procedure **a**, (catalyst / substrate ratio - 1:12), yield: 51%, as an oil: $[\alpha]_D -15.9$ (*c* 3.8, chloroform); IR ν_{\max} 2960, 1640, 1225, 1105, 1040 cm^{-1} ; $^1\text{H NMR}$ δ 5.39 (m, 1H, H-10'), 4.68 (d, $J_{1,2}$ 6.7, 1H, H-1), 3.95-3.65 (m, 6H, H-3, 4, 5, 6, 6', 1'), 3.58 (dd, $J_{2,3} < 1.0$, 1H, H-2), 3.37 (ddd, $2 \times J_{1'',2'}$ 6.5, $J_{1',1''}$ 9.0, 1H, H-1'), 1.97 (m, 2H, H-9'), 1.61 (m, 2H, H-2'), 1.29 (m, 12H, CH₂ alkyl), 0.89-0.87 (4xs, 36H, 4x(CH₃)₃CSi), 0.09-0.05 (8xs, 24H, 8xCH₃Si); $^{13}\text{C NMR}$ δ : 130.3, 129.9 (*cis* and *trans* C-10'), 101.7 (C-1), 82.3 (C-3), 79.2 (C-2), 77.8 (C-4), 70.1 (C-5), 69.5 (C-1'), 64.1 (C-6), 32.6-26.3 (8xC alkyl), 26.1-25.8 (12xCH₃ *t*-butyl), 18.3-17.9 (4xSiCMe₃), -4.1--5.3 (8xSiCH₃); FABMS: 797.9 (M+2Na)⁺⁺, 863.6, 848.6, 834.8, 811.7.

Anal. Calcd for C₈₀H₁₇₂O₁₂ Si₈: C, 61.96; H, 11.18. Found: C, 62.67; H, 10.75.

The same product **7** was obtained by procedure **b** (catalyst / substrate ratio - 1:12), yield: 92%.

(E,Z) Di-O-(2,3,4,6-tetra-O-benzyl- β -D-glucopyranosyloxy)1,20-eicos-10-ene 8. The product was obtained from substrate **5** by procedure **a**, (catalyst / substrate ratio - 1:5), yield: 37%, as an oil, $[\alpha]_D -15.9$ (*c* 0.85, chloroform); IR ν_{\max}

2920, 1640, 1630, 1585, 1225, 1040 cm^{-1} ; ^1H NMR δ 7.4-7.2 (m, 20H, arom), 5.37 (bt, 1H, H-10'), 4.84, 4.82, 4.67, 5.58 (4 x ABq, J_{AB} 11.0, 11.0, 10.8, 12.3, 8H, CH_2 benzyl), 4.38 (d, $J_{1,2}$ 7.7, 1H, H-1), 3.95 (ddd, $2xJ_{1',2'}$ 6.3, $J_{1'',1''}$ 9.3, 1H, H-1'), 3.74 (dd, $J_{5,6}$ 2.0, $J_{6,6'}$ 10.9, 1H, H-6), 3.72-3.39 (m, 6H, H-2, 3, 4, 5, 6', 1''), 2.10 (m, 2H, H-9'), 1.65 (m, 2H, H-2'), 1.29 (m, 12H, CH_2 alkyl); ^{13}C NMR: 138.7-138.2 (4xC arom), 130.4 (C-10'), 128.4-127.6 (C arom), 103.7 (C-1), 84.8 (C-3), 82.3 (C-2), 78.0 (C-4), 75.7, 75.0, 74.7, 73.5 (4x CH_2 benzyl), 74.8 (C-5), 70.2 (C-1'), 69.0 (C-6), 32.6-26.2 (C alkyl).

Anal. Calcd for $\text{C}_{88}\text{H}_{108}\text{O}_{12}$ Sig: C, 77.84; H, 8.02. Found: C, 78.02; H, 7.78.

The same product **8** was not obtained by procedure **b**, (catalyst / substrate ratio - 1:15),

(E,Z) Di-O-(2-acetamido-3,4,6-tri-O-acetyl-2-deoxy- β -D-glucopyranosyloxy)-1,20-eicos-10-ene 12. The product was obtained from substrate **9** by procedure **a**, (catalyst / substrate ratio - 1:10), yield: 62%, as an oil: $[\alpha]_{\text{D}} -13.5$ (*c* 1.3, chloroform); IR ν_{max} 2920, 1760, 1640, 1240, 1050 cm^{-1} ; ^1H NMR δ 5.63 (bd, $J_{1,\text{NH}}$ 8.6, 1H, NH), 5.37 (bt, 1H, H-10'), 5.29 (dd, $J_{3,4}$ 9.0, 1H, H-3), 5.07 (dd, $J_{4,5}$ 9.5, 1H, H-4), 4.69 (d, $J_{1,2}$ 8.3, 1H, H-1), 4.28 (dd, $J_{5,6}$ 4.6, 1H, H-6), 4.13 (dd, $J_{6,6'}$ 12.3, 1H, H-6'), 3.86 (ddd, $J_{2,3}$ 9.5, 1H, H-2), 3.84 (ddd, $J_{1',1''}$ 9.6, $2xJ_{1',2'}$ 6.4, 1H, H-1'), 3.71 (ddd, $J_{5,6'}$ 2.1, 1H, H-5), 3.48 (ddd, $2xJ_{1'',2''}$ 6.4, 1H, H-1''), 2.08, 2.03, 2.02, 1.94 (4xs, 12H, 4x CH_3CO), 1.97 (m, 2H, H-9'), 1.56 (m, 2H, H-2'), 1.26 (m, 12H, CH_2 alkyl); ^{13}C NMR 170.8, 170.7, 170.1, 169.4 (4 x CO), 130., 129.8 (*cis* and *trans* C-10'), 100.6 (C-1), 72.3 (C-3), 71.7 (C-5), 69.9 (C-1'), 68.7 (C-4), 62.2 (C-6), 54.8 (C-2), 32.5-25.8 (8xC alkyl), 23.3 (CH_3 amide), 20.7-20.6 (3 x CH_3 CO). FABMS: 994.8 (M+Na)⁺, 972.4 (M+H)⁺, 951.1, 330.3.

Anal. Calcd for $\text{C}_{48}\text{H}_{78}\text{O}_{18}$ N₂: C, 59.37; H, 8.10; N, 2.88. Found: C, 59.29; H, 8.14; N, 2.74.

(E,Z) Di-O-(2,3,4,6-tetra-O-acetyl- β -D-glucopyranosyloxy)-1,8-oct-4-ene 13. The product obtained from substrate **10** by procedure **a**, (catalyst / substrate ratio - 1:20), yield: 65%, as an oil, $[\alpha]_{\text{D}} -17.5$ (*c* 1.0, chloroform); IR ν_{max} 2950, 1745, 1620, 1220, 1040 cm^{-1} ; ^1H NMR δ 5.38 (m, 1H, H-4'), 5.18 (dd, $J_{3,4}$ 9.5, 1H, H-3), 5.10 (dd, $J_{4,5}$ 9.3, 1H, H-4), 4.98 (dd, $J_{2,3}$ 9.2, 1H, H-2), 4.49 (d, $J_{1,2}$ 7.9, 1H, H-1), 4.27 (dd, $J_{5,6}$ 4.6, 1H, H-6), 4.13 (dd, $J_{6,6'}$ 12.3, 1H, H-6'), 3.86 (ddd, $J_{1',1''}$ 9.6, $2xJ_{1',2'}$ 6.2, 1H, H-1'), 3.69 (ddd, $J_{5,6'}$ 2.4, 1H, H-5), 3.48 (ddd, $2xJ_{1'',2''}$ 6.5, 1H, H-1''), 2.08, 20.4, 20.2, 20.1 (4xs, 12H, 4x CH_3CO), 2.05 (m, 2H, H-3'), 1.62 (m, 2H, H-2'); ^{13}C NMR 170.6, 170.2, 169.4, 169.2 (4 x CO), 129.9, 129.4 (*cis* and *trans* C-4'), 100.7 (C-1), 72.8 (C-3), 71.7 (C-5), 71.3 (C-2), 69.4 (C-1'), 68.4 (C-4), 61.9 (C-

6), 29.1, 28.5 (2xCH₂ alkyl), 20.7-20.6 (4 x CH₃ CO). FABMS: 827.3 (M+Na)⁺, 805.4 (M+H)⁺, 518.3, 331.6.

Anal. Calcd for C₃₆H₅₂O₂₀: C, 53.73; H, 6.51. Found: C, 53.51; H, 6.52.

No metathesis product was obtained from allyl 2,3,4,6-tetra-*O*-acetyl- β -D-glucopyranoside **11** either by the procedure **a** or the procedure **b**.

(E,Z)-Di-O-(4,6-di-O-acetyl-2,3-dideoxy- α -D-erythrohex-2-eno-pyranosyloxy)-1,8-oct-4-ene 16. The product was obtained from substrate **14** by procedure **a**, (catalyst / substrate ratio - 1:20), yield: 52%, as an oil, $[\alpha]_D +100.0$ (*c* 0.2, chloroform); IR ν_{\max} 2920, 1750, 1640, 1225, 1040 cm⁻¹; ¹H NMR δ 5.87 (m, 2H, H-2, H-3), 5.42 (bt, 1H, H-4'), 5.32 (dd, J_{5,4} 9.4, J_{3,4} 0.8, 1H, H-4), 5.02 (bs, J_{1,2}=J_{1,3} 1.6, 1H, H-1), 4.23 (dd, J_{5,6} 5.2, 1H, H-6), 4.18 (dd, J_{6,6'} 12.1, J_{5,6'} 2.3, 1H, H-6'), 4.09 (ddd, 1H, H-5), 3.79 (ddd, J_{1',1''} 9.6, 2xJ_{1',2'} 6.5, 1H, H-1'), 3.50 (ddd, 2xJ_{1'',2'} 6.5, 1H, H-1''), 2.15 (m, 2H, H-3'), 2.10, 2.08 (2xs, 6H, 2xCH₃CO), 1.66 (m, 2H, H-2'); ¹³C NMR: 170.8, 170.3 (2x CO), 129.5, 129.0 (*cis* and *trans* C-4'), 130.0 (C-3), 127.9 (C-2), 94.5 (C-1), 68.3 (C-1'), 66.8 (C-5), 65.2 (C-4), 63.0 (C-6), 29.6, 29.2 (2xCH₂ alkyl), 21.0, 20.8 (2 x CH₃ CO).

Anal. Calcd for C₂₈H₄₀O₁₂: C, 59.14; H, 7.09. Found: C, 58.92; H, 7.16.

(E,Z) Di-O-(4,6-di-O-acetyl-2,3-dideoxy- α -D-erythro-2-eno-pyranosyloxy)-1,10-dec-5-ene 17. The product was obtained from substrate **15** by procedure **a**, (catalyst / substrate ratio - 1:27), yield: 55%, as an oil, $[\alpha]_D +88.3$ (*c* 1.8, chloroform); IR ν_{\max} 2920, 1750, 1640, 1225, 1040 cm⁻¹; ¹H NMR δ 5.83 (m, 2H, H-2, H-3), 5.39 (bt, 1H, H-5'), 5.30 (dd, J_{5,4} 9.4, J_{3,4} 0.8, 1H, H-4), 5.01 (bs, J_{1,2}=J_{1,3} 1.6, 1H, H-1), 4.25 (dd, J_{5,6} 5.2, 1H, H-6), 4.18 (dd, J_{6,6'} 12.1, J_{5,6'} 2.3, 1H, H-6'), 4.10 (ddd, 1H, H-5), 3.79 (ddd, J_{1',1''} 9.6, 2xJ_{1',2'} 6.5, 1H, H-1'), 3.50 (ddd, 2xJ_{1'',2'} 6.5, 1H, H-1''), 2.10, 2.08 (2xs, 6H, 2xCH₃CO), 2.02 (m, 2H H-4'), 1.60 (m, 2H, H-2'), 1.40 (m, 2H, H-3'); ¹³C NMR: 169.7, 169.2 (2x CO), 129.5 (C-3), 129.2, 129.0 (*cis* and *trans* C-5'), 126.8 (C-2), 93.3 (C-1), 67.7 (C-5), 65.8 (C-1'), 64.2 (C-4), 61.9 (C-6), 31.2, 28.2 (2xCH₂ alkyl), 19.9, 19.7 (2 x CH₃ CO).

Anal. Calcd for C₃₀H₄₄O₁₂: C, 60.39; H, 7.43. Found: C, 60.46; H, 7.53.

(E,Z) Di-O-(β -D-glucopyranosyloxy)-1,20-eicos-10-ene 18. To a solution of compound **6** (100 mg, 0.10 mM) in methanol (5 mL) a catalytic amount of MeONa was added and the resulting mixture was stirred overnight. After neutralisation (acetic acid) and solvent evaporation, the residue was filtered through a short column (CH₂Cl₂/MeOH 2:1) affording **18** as an oil (52%), $[\alpha]_D +26.7$ (*c* 0.06, methanol); IR ν_{\max} 3500, 2920, 1225, 1040 cm⁻¹; ¹H NMR (CD₃OD) δ 5.37 (m, 1H, H-10'), 4.24 (d, J_{1,2} 7.7, 1H, H-1), 3.97-3.12 (m, 8H, H-2,3, 4, 5, 6, 6', 1', 1''), 2.02 (m, 2H, H-9'), 1.65 (m, 2H, H-2'), 1.31 (m, 12H, CH₂ alkyl); ¹³C NMR (CD₃OD) δ : 131.5, 129.9 (*cis*

and *trans* C-10'), 104.4 (C-1), 78.1 (C-3), 77.9 (C-5), 75.1 (C-2), 71.7 (C-4), 70.9 (C-1'), 62.8 (C-6), 33.6-27.1 (8xC_{alkyl}); FABMS: 659.4 (M+Na)⁺, 429.3, 413.4, 409.4, 393.4, 351.4, 317.3.

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